

## NEW DIESTERS OF 12-DEOXY-PHORBOL

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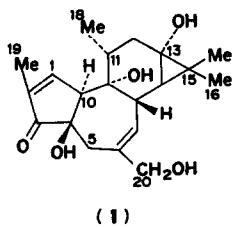
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**Key Word Index**—*Euphorbia* species; Euphorbiaceae; 12-deoxyphorbol esters; diterpene.

12-Deoxy-phorbol esters are potent toxins which produce severe inflammation of the skin [1]. They are therefore of interest in the study of the complex biochemical sequences involved in the inflammatory process in mammals [2]. Their occurrence in *Euphorbia* species is of particular interest in this respect due to a recent failure to synthesize the 12-deoxy-phorbol series of irritants from the phorbol series [3]. We wish to report the isolation of three new irritant diesters of 12-deoxy-4 $\beta$ OH-phorbol which occurred as only minor components of a complex mixture of known esters [2,4] from the latex of *Euphorbia* species.

*Plant material.* *E. coerulescens*; *E. fortissima*; *E. polyacantha*.

*Present work.* Approximately 10 ml fresh latex was used for the isolations. The irritants were extracted and purified by a combination of solvent partition, column chromatography and TLC as previously described [4]. Final purification was by preparative TLC on Si gel (500  $\mu$ m layers activated at 120° for 1 hr after buffering at pH 7.0) developing with  $\text{CHCl}_3\text{-C}_6\text{H}_6\text{-Et}_2\text{O}$  (1:3:3) as solvent. Clear glassy resins were produced which failed to crystallize from several solvents, but which were homogeneous by TLC [5] and MS.



The resins had similar IR spectra ( $\nu_{\text{max}}$ , 3385, 1725, 1705, 1695 and 1630  $\text{cm}^{-1}$ , solvent  $\text{CHCl}_3$ )

and CD spectra (negative cotton effects at 272 and 360 nm, solvent MeOH). Hydrolysis [ $\text{Ba(OH)}_2$  in MeOH under  $\text{N}_2$ ] afforded a common parent alcohol (1) in *ca* 40% yield. This alcohol was unstable and was converted to its diacetate [6] before recrystallising from  $\text{Me}_2\text{CO}$  (mp 138°). The NMR spectrum (60 MHz,  $\text{CDCl}_3$ ; TMS = 0.00 ppm) exhibited signals at  $\delta$  0.89 *d* (*J* 4 Hz, 3H-18, H-14);  $\delta$  1.16 *d* (*J* 8 Hz, 6H-16, 17);  $\delta$  1.80 *d* (*J* 2.1 Hz, 3H-19);  $\delta$  2.05 (6H-Me-CO-);  $\delta$  2.18 *s* (2H-12);  $\delta$  2.44 *s* (2H-5);  $\delta$  3.02 *m* (H-8);  $\delta$  3.29 *m* (H-10);  $\delta$  4.47 *s* (2H-20);  $\delta$  5.75 *d* (*J* 4.3 Hz, H-7);  $\delta$  7.64 *s* (H-1);  $\delta$  2.47 and  $\delta$  5.58 (2-OH-deuterium exchange). The MS exhibited a molecular ion at *m/e* 432 ( $\text{M}^+ \text{C}_{24}\text{H}_{32}\text{O}_7$ ) with significant fragment ions at *m/e* 414 (M-18); 372 (M-60); 354 (M-60 + 18); 336 (M-36 + 60); 312 (M-120); 294 (M-120 + 18) in the upper region of the spectrum. The TLC [5] and GLC [6] data were identical to 12-deoxy-4 $\beta$ OH-phorbol-13,20-diacetate.

*Ester A:* 12-deoxy-4 $\beta$ OH-phorbol-13-dodecanoate-20-acetate. This resin (10 mg) was isolated from latex of *E. coerulescens*. It had an *R<sub>f</sub>* value of 0.8 in the system above, and produced an orange fluorescence under UV light after spraying with 60%  $\text{H}_2\text{SO}_4$  and heating. Most of the signals in the NMR spectrum could be characterized as arising from the parent alcohol [7]. In addition, it had 3H signal at  $\delta$  0.89 and an 18H, *s* at  $\delta$  1.27, a 2H triplet at  $\delta$  2.32 and one acetate at  $\delta$  2.05. The MS *m/e* 572 ( $\text{M}^+ \text{C}_{34}\text{H}_{52}\text{O}_2$ ) and prominent fragment ions at *m/e* 512 (M-60); 372 (M-200); 312 (M-60 + 200); 294 (M-60 + 200 + 18). Transesterification in 0.5 M KOH in MeOH at room temp. produced a low *R<sub>f</sub>* value mono-ester. [ $\text{M}^+ \text{C}_{32}\text{H}_{50}\text{O}_6$  at *m/e* 530, fragment ions by MS at *m/e* 512 (M-18); 494 (M-36); 330 (M-

200); 312 (M-200 + 18).] Acetylation of the mono-ester produced a compound identical by TLC and MS to *ester A*. Hydrolysis in Ba(OH)<sub>2</sub> followed by methylation and GLC[8] confirmed the presence of dodecanoic acid.

*Ester B:* 12-deoxy-4 $\beta$ OH-phorbol-13-dodecanoate-20-acetate. Resin B (5 mg:  $R_f$  value 0.75, orange by UV as before) was isolated from *E. fortissima* latex. The NMR spectrum as before suggested the presence of acetic and dodecanoic acids as esterifying moieties at C-20 and C-13 of (1). In the MS the resin had an  $M^+$  ion at  $m/e$  570 ( $M^+ C_{34}H_{50}O_7$ ) and fragment ions at  $m/e$  510 (M-60); 494 (M-60 + 18); 372 (M-198); 312 (M-198 + 60); 294 (M-198 + 60 + 18). Transesterification produced a mono-ester [ $M^+ C_{32}H_{48}O_6$ , at  $m/e$  528 and fragment ions by MS at  $m/e$  510 (M-18); 492 (M-36); 330 (M-198); 312 (M-198 + 18)]. Acetylation of the mono-ester produced *ester B*. After complete hydrolysis dodecanoic acid was identified by GLC as before.

*Ester C:* 12-deoxy-4 $\beta$ OH-phorbol-13-octenoate-20-acetate. This ester (1.5 mg) was isolated from *E. polyacantha* ( $R_f$  value 0.72, orange by UV as before). It exhibited a molecular ion in the MS at  $m/e$  514 ( $M^+ C_{30}H_{42}O_7$ ) and fragment ions at  $m/e$  372 (M-142);  $m/e$  454 (M-60);  $m/e$  312 (M-60 + 142);  $m/e$  294 (M-60 + 142 + 18). Transesterification produced a mono-ester. (MS exhibited

$M^+$  at  $m/e$  472,  $C_{28}H_{40}O_6$ , and fragment ions at  $m/e$  454 (M-18); 436 (M-36); 330 (M-142). Octenoic acid was identified by GLC after hydrolysis. Acetylation of the mono-ester produced *ester C*.

For *esters B* and *C* no attempt was made to assign the position of the double bond in the side chain. From a chemotaxonomic point of view it was of interest to note that these three succulent *Euphorbia* species, which are indigenous to Africa, all contained esters of the same parent alcohol (1).

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## DITERPENES FROM THREE *SIDERITIS* SPECIES\*

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**Key Word Index**—*Sideritis chamaedryfolia*; *S. hyssopifolia*; *S. luteola*; Labiate; *ent*-15-kaurene and *ent*-16-kaurene derivatives.

**Plants.** *Sideritis chamaedryfolia* Cav., *Sideritis hyssopifolia* L. and *Sideritis luteola* Font Quer.  
**Sources.** Near Villena (Alicante), Puerto de Pajares (León) and Sierra de Filabres (Almería),

\* Part 24 in the series *Constituents of Sideritis*. For part 23 see Von Carstenn-Lichterfelde, C., Panizo, F. M., Quesada, T. G., Rodriguez, B., Valverde, S., Ayer, W. A. and Ball, J.-A. H. *Can. J. Chem.* (in press).